

THREO- AND ERYTHRO- $\beta$ -HYDROXY-L-ASPARAGINES

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(Received in UK 5 August 1968; accepted for publication 12 August 1968)

The recent findings of Campbell et al. (1) regarding the action of L-asparaginase on various asparagine-requiring tumours have prompted the syntheses and investigation of analogues of this amino acid (2).

We wish to report the preparation of the  $\beta$ -amides of threo- and erythro- $\beta$ -hydroxy-L-aspartic acids by ammonolysis of  $\beta$ -benzyl threo- $\beta$ -hydroxy-L-aspartate (3) and  $\beta$ -methyl erythro- $\beta$ -hydroxy-L-aspartate respectively. The latter ester was prepared by direct esterification of erythro- $\beta$ -hydroxy-L-aspartic acid obtained by resolution of its racemic N-benzyl derivative by means of L-histidine and subsequent hydrogenolysis (4).

The amides differ in their stability in acidic medium. The one belonging to the threo- series when left for several hours in N HCl is partly hydrolysed to threo- $\beta$ -hydroxy-L-aspartic acid, detected on chromatograms, which caused a sharp drop in optical rotation. Erythro- $\beta$ -hydroxy-L-asparagine, however, was stable under the same conditions.

The " $\beta$ -hydroxy asparagine" isolated from normal human urine (5) could not be identified with either of the isomers synthesised by us, having different properties ( $R_f$  values and IR spectrum).

Threo- $\beta$ -hydroxy-L-asparagine -  $\beta$ -Benzyl threo- $\beta$ -hydroxy-L-aspartate (3) (1 g) was dissolved in 25% aqueous ammonia (10 ml). After 7 days at room temperature the solution was evaporated to dryness in vacuo. The solid residue was dissolved in hot water and the pH adjusted to 5 by addition of 6N HCl. The substance precipitated and was recrystallised from water (0.6 g, yield quantitative); m.p. 278° (decomp.);  $[\alpha]_D^{22}$  -30.6° (c 2.45 in N HCl) (Found: C, 32.2; H, 5.9; N, 18.9; N (Van Slyke), 9.6.  $C_4H_8N_2O_4$  requires: C, 32.4; H, 5.4; N, 18.9; N (Van Slyke), 9.4%).

$\beta$ -Methyl erythro- $\beta$ -hydroxy-L-aspartate - To a suspension of erythro- $\beta$ -hydroxy-L-aspartic acid (4) (5 g) in anhydrous methanol (50 ml) was added concentrated HCl (6 ml). The mixture was heated under reflux for 3 hrs. The

solvents were removed in vacuo, the residue was dissolved in ethanol and the pH adjusted to 8 by addition of pyridine. The resulting precipitate was recrystallised from 50% aqueous ethanol. (3.5 g, 65%); m.p. 229° (decomp.);  $[\alpha]_D^{22} +65.5^\circ$  (c, 1.8 in N HCl) (Found: C, 36.8; H, 5.8; N, 8.5; N (Van Slyke), 8.7.  $C_5H_9NO_5$  requires: C, 36.8; H, 5.5; N, 8.6; N (Van Slyke), 8.6%).

Erythro- $\beta$ -hydroxy-L-asparagine was prepared in an analogous manner to the threo isomer by ammonolysis of the above ester; m.p. 260° (decomp.);  $[\alpha]_D^{22} +63.4^\circ$  (c 1.9 in N HCl) (Found: C, 32.4; H, 5.9; N, 18.9; N (Van Slyke), 9.2.  $C_4H_8N_2O_4$  requires: C, 32.4; H, 5.4; N, 18.9; N (Van Slyke), 9.4%).

Paper chromatography in 80% phenol-water (light-brown colour with ninhydrin) gave  $R_f$  values 0.18 and 0.13 for threo- and erythro- $\beta$ -hydroxy-L-asparagine respectively. On TLC (silica gel) in n-butanol-acetic acid-pyridine-water (2:1:1:4) they were 0.39 and 0.23 respectively.

#### REFERENCES

- 1) H.A. Campbell, L.T. Mashburn, E.A. Boyse and L.J. Old, Biochemistry, **6**, 721 (1967).
- 2) E. Falco and G.B. Brown, J. Medicinal Chem., **11**, 142 (1968).
- 3) Y. Liwschitz, A. Singerman and S. Sokoloff, J. Chem. Soc. (C), 1968 in press.
- 4) Y. Liwschitz, A. Singerman and Y. Wiesel, Israel J. Chem., 1968 in press.
- 5) F. Tominaga, C. Hiwaki, T. Maekawa and H. Yoshida, J. Biochem. (Japan) **52**, 227 (1963).